A metal plate of area 1 × 10⁻⁴ m² is illuminated by a radiation of intensity 16 mW/m2. The work function of the metal is 5 eV. The energy of the incident photons is 10 eV and only 10% of it produces photo electrons. The number of emitted photo electrons per second and their maximum energy, respectively, will be:

[1 eV = 1.6×10^{-19} J] [Main 10 Jan. 2019 (II)] (a) 10^{14} and 10 eV (b) 10^{12} and 5 eV

(c) 10¹¹ and 5 eV (d) 10¹⁰ and 5 eV

ANS (c) using, intensity $I = \frac{nE}{I}$

n = no. of photoelectrons

$$\Rightarrow 16 \times 10^{-3} = \left(\frac{n}{t}\right) \times \frac{10 \times 1.6 \times 10^{-19}}{10^{-4}} \text{ or, } \frac{n}{t} = 10^{12}$$

So, effective number of photoelectrons ejected per unit time = $10^{12} \times 10/100 = 10^{11}$