

Important concept

To be remembered:

Order of strength of intermolecular forces from strongest to weakest

S.no	Type of interaction	Interacting species
1	Ion-Ion	Ions only
2	Ion-dipole	Ions and polar molecules
3	Hydrogen bonding	N,O,F linked to H atom
4	Dipole-dipole	Polar molecules
5	Dipole-induced dipole	Atleast one molecule should be polar
6	London dispersion forces	All types of molecules

- Try to gather more information on each type of interaction