Arrange following complex ions in increasing order of crystal field splitting energy (Δ_{o}) :

[Cr(Cl)₆]³⁻, [Cr(CN)₆]³⁻, [Cr(NH₃)₆]³⁺.

As CFSE depends on ligand Field strength the following is the order :

Crystal field splitting energy increases in the order [Cr(Cl) $_6$] $^{3-}$ < [Cr(NH $_3)_6$] $^{3+}$ < [Cr(CN) $_6$] $^{3-}$