Three moles of ${\rm B_2H_6}$ are completely reacted with methanol. The number of moles of boron containing product formed is (2015 Adv.)

 $\begin{array}{ll} B_2H_6+6 & CH_3OH \longrightarrow 2[B(OCH_3)_3]+6H_2 \\ \\ Therefore, from 3 moles of B_2H_6, 6 moles of B(OCH_3)_3 will be \end{array}$