- 4. Which of the following arrangements represent increasing oxidation number of the central atom?
- (i) CrO<sup>2-</sup>, CIO<sup>-3</sup>, CrO2<sup>-4</sup>, MnO<sup>-4</sup>
- (ii) CIO<sup>3</sup>, CrO2<sup>4</sup>, MnO<sup>4</sup>, CrO<sup>2</sup>
- (iii)  $CrO2^{{\scriptscriptstyle +4}}$  ,  $MnO^{{\scriptscriptstyle -4}}$  ,  $CrO^{{\scriptscriptstyle -2}}$  ,  $CIO^{{\scriptscriptstyle 3}{\scriptscriptstyle -}}$
- (iv)  $CrO2^{4-}$ ,  $MnO^{4-}$ ,  $CrO^{2-}$ ,  $CIO^{3-}$

## **Solution:**

Option (i) is the answer