- The bulk moduli of ethanol, mercury and water are given as 0.9, 25 and 2.2 respectively in units of 10^9 Nm⁻². For a given value of pressure, the fractional compression in volume is $\frac{\Delta V}{V}$. Which of the following statements about
- volume is $\frac{\Delta V}{V}$. Which of the following statements about $\frac{\Delta V}{V}$ for these three liquids is correct ?[Online April 11, 2014]
 - V February 12 12 12 14 15 14
 - (a) Ethanol > Water > Mercury
 - (b) Water > Ethanol > Mercury
- (c) Mercury > Ethanol > Water(d) Ethanol > Mercury > Water

27. (a) Compressibility= $\frac{1}{\text{Bulk modulus}}$ As bulk modulus is least for ethanol (0.9) and maximum for

mercury (25) among ehtanol, mercury and water. Hence compression in volume $\frac{\Delta V}{V}$

Ethanol > Water > Mercury