

When Γ^- is oxidised by MnO_4^- in alkaline medium,
 Γ^- converts into (2004)

- (a) IO_3^- (b) I_2 (c) IO_4^- (d) IO^-



When MnO_2 is fused with KOH , a coloured compound is
formed, the product and its colour is (2003, 1M)

- (a) K_2MnO_4 , purple green (b) KMnO_4 , purple
(c) Mn_2O_3 , brown (d) Mn_3O_4 , black

Ammonium dichromate on heating produces $\text{N}_2(\text{g})$. NH_4NO_2
also gives N_2 on heating :

