

8. The pressure-volume work for an ideal gas can be calculated by using the

expression  $w = - \int_{V_i}^{V_f} p_{ex} dV$ . The work can also be calculated from the  $pV$ -plot

by using the area under the curve within the specified limits. When an ideal gas is compressed (a) reversibly or (b) irreversibly from volume  $V_i$  to  $V_f$ , choose the correct option.

- (i)  $w$  (reversible) =  $w$  (irreversible)
- (ii)  $w$  (reversible) <  $w$  (irreversible)
- (iii)  $w$  (reversible) >  $w$  (irreversible)
- (iv)  $w$  (reversible) =  $w$  (irreversible) +  $p_{ex} \cdot \Delta V$

Solution:

(ii)  $w$ (reversible) <  $w$  (irreversible)

Explanation:

