

Q4: According to Einstein's photoelectric equation, the plot of the kinetic energy of the emitted photoelectrons from a metal vs the frequency of the incident radiation gives a straight line whose slope

- (a) depends on the nature of the metal used
- (b) depends on the intensity of the radiation
- (c) depends both on the intensity of the radiation and the metal used
- (d) is the same for all metals and independent of the intensity of the radiation.

Solution:

According to photoelectric equation:

$$K = hf - \Phi$$

Where, f = frequency of light used

Φ = work function of metal

K = kinetic energy

On comparing with $y = mx + c$

Slope = $m = h$ = planck's constant

Which is independent of nature of metal used or light