

## Previous Year JEE Problems with Explanations

What is the work function of the metal if the light of wavelength  $4000\text{Å}$  generates photoelectrons of velocity  $6 \times 10^5 \text{ms}^{-1}$  from it ?

(Mass of electron =  $9 \times 10^{-31} \text{kg}$ ;

Velocity of light =  $3 \times 10^8 \text{ms}^{-1}$

Plank's constant =  $6.626 \times 10^{-34} \text{Js}$ ;

Charge of electron =  $1.6 \times 10^{-19} \text{JeV}^{-1}$ )

- A** 4.0 eV
- B** 0.9 eV
- C** 2.1 eV
- D** 3.1 eV

### Explanation

$$E = \phi + \text{K.E}$$

$$\Rightarrow h\nu = \phi + \frac{1}{2}mv^2$$

$$\Rightarrow \phi = h\nu - \frac{1}{2}mv^2$$

$$= \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{4000 \times 10^{-10}} - \frac{1}{2} \times 9 \times 10^{-31} \times (6 \times 10^5)^2$$

$$= 3.35 \times 10^{-19} \text{J}$$

$$\Rightarrow \phi = \frac{3.35 \times 10^{-19}}{1.6 \times 10^{-19}} \text{eV}$$

$$= 2.0934 \text{eV} \simeq 2.1 \text{eV}$$

