PREVIOUS YEAR QUESTION

For any given series of spectral lines of atomic hydrogen, let $\Delta \bar{v} = \Delta \overline{v}_{\max} - \Delta \overline{v}_{\min}$ be the difference in maximum and minimum frequencies in cm⁻¹. The ratio Lyman Balmer $\frac{\Delta \bar{v}_{Lyman}}{\Delta \bar{v}_{Balmer}}$ is :

- A 9:4
- **B** 4:1
- C 27:5
- 5:4

Explanation

We know,

$$\overline{v} = RZ^2 \left(rac{1}{n_1^2} - rac{1}{n_2^2}
ight)$$

∴ For Lyman series,

$$\Delta \overline{v}_{Lyman}$$
 = $\Delta \overline{v}_{max} - \Delta \overline{v}_{min}$

$$= \left[\frac{1}{1} - \frac{1}{\infty}\right] - \left[\frac{1}{1} - \frac{1}{4}\right]$$

$$=\frac{1}{4}$$

∴ For Balmer series,

$$\Delta \overline{v}_{Balmer}$$
 = $\Delta \overline{v}_{ ext{max}} - \Delta \overline{v}_{ ext{min}}$

$$= \left[\frac{1}{4} - \frac{1}{\infty}\right] - \left[\frac{1}{4} - \frac{1}{9}\right]$$

$$=\frac{1}{9}$$

$$\therefore \frac{\Delta \overline{v}_{Lyman}}{\Delta \overline{v}_{Bulmer}} = \frac{\frac{1}{4}}{\frac{1}{4}} = \frac{9}{4}$$