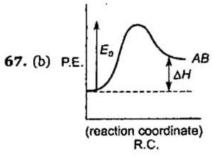
- 67. For an endothermic reaction where ΔH represents the enthalpy of reaction in kJ/mol, the minimum value for the energy of activation will be:
 - (a) less than ΔH
- (b) more than ΔH
- (c) equal to ΔH
- (d) zero

ANSWER:



 E_e is always greater than ΔH