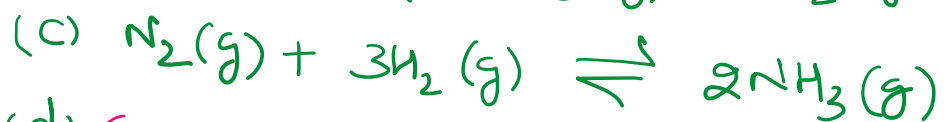
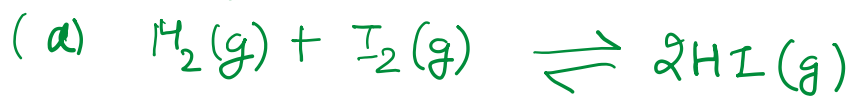


In which of the following cases equilibrium remains unaffected by adding small amount of Argon at constant volume??



(d) ~~The equilibrium will remain unaffected in all cases.~~

Sol. Adding small amount of inert gas at constant volume doesn't affect any equilibrium.

So, option (d) is correct.