$B_2H_6 + NaH \longrightarrow ?$

The correct statement(s) is / are

- (A) Symmetrical cleavage takes place

(B) NaBH₄ is produced

(C) H₂ is produced

(D) Both (A) and (B)

Sodium hydride react with diberane.

2 NaH + B2 H4 -> 2 Na[BH4]