

Which statement is correct?

(A) Tl^{3+} salts act as good oxidising agents

(B) In borax bead test the coloured ions give characteristic coloured bead due to formation of metal phosphates

(C) BCl_3 and $AlCl_3$ both are lewis acids but $AlCl_3$ is stronger lewis acid than BCl_3

(D) Boric acid is a protonic acid

Ans : A

Tl^{+3} salts are very less often formed as Tl^{+3} is unstable