The number of 2-centre-2-electron and 3-centre-2-electron bonds in $\rm B_2H_6$, respectively, are $\,$ (2019 Main, 10 Jan II)

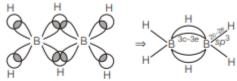
(a) 4 and 2

(b) 2 and 4

(c) 2 and 2

(d) 2 and 1

The structure of B2H6 can be shown as :



In B_2H_6 , four 2-centre-2-electron (2c-2e) bonds are present in the same plane and two 3-centre-2-electron (3c-2e) bonds are present in another plane.