Identify the compounds A, \boldsymbol{X} and \boldsymbol{Z} in the following reactions :

(i)
$$\begin{array}{ccc} A + 2HCl + 5H_{2}O & \longrightarrow 2NaCl + X \\ & X & \xrightarrow{\Delta} & HBO_{2} & \xrightarrow{\Delta} & Z \end{array}$$

 $\begin{array}{lll} A & : & \operatorname{Na_2B_4O_7} \operatorname{(Borax)} \\ X & : & \operatorname{H_3BO_3} \\ Z & : & \operatorname{B_2O_3} \end{array}$