$\mathbb{Q}.54$ A primary amine, RNH₂ can be reacted with CH₃—X to get secondary amine, R—NHCH₃ but the only disadvantage is that 3° amine and quaternary ammonium salts are also obtained as side products. Can you suggest a method where RNH2 forms only 2° amine?

 $\begin{array}{ccc} RNH_2 & \xrightarrow{KOH/CHCl_3} & RNC \\ Primary & amine \end{array} \xrightarrow{KOH/CHCl_3} & RNC \\ & & & & & & & & & & & & & \\ Primary & amine & & & & & & & & & & \\ \hline Primary & amine & & & & & & & & & & \\ \hline Primary & amine & & & & & & & & & \\ \hline Primary & amine & & & & & & & & & \\ \hline Primary & amine & & & & & & & & \\ \hline Primary & amine & & & & & & & \\ \hline Primary & amine & & & & & & & \\ \hline Primary & amine & & & & & & \\ \hline Primary & amine & & & & & & \\ \hline Primary & amine & & & & & \\ \hline Primary & amine & & & & & \\ \hline Primary & amine & & & & \\ \hline Primary & amine & & & & \\ \hline Primary & amine & & & & \\ \hline Primary & amine & & & & \\ \hline Primary & amine & & & & \\ \hline Primary & amine & & \\ \hline Primary & ami$ Secondary amine Primary amines show carbylamine reaction in which two H-atoms attached to N-atoms of NH₂ are replaced by one C-atom. On catalytic reduction, isocyanide (formed) produces

secondary amine and not tertiary or quaternary salts.

Ans.