- In the Hofmann-bromamide degradation reaction, the number of moles of NaOH and Br₂ used per mole of amine produced are (2016 Main)
- (a) four moles of NaOH and two moles of Br₂
- (b) two moles of NaOH and two moles of Br_2
- (c) four moles of NaOH and one mole of Br_2
- (d) one mole of NaOH and one mole of Br₂

16. Hofmann-bromamide degradation reaction is given as: $RCONH_2 + 4NaOH + Br_2 \longrightarrow RNH_2$

+ Na₂CO₃ + 2NaBr + 2H₂O

Hence, four moles of NaOH and one mole of Br₂ are used.

(1° amine)