

7. 10 mL of 1 mM surfactant solution forms a monolayer covering  $0.24 \text{ cm}^2$  on a polar substrate. If the polar head is approximated as a cube, what is its edge length?

[April 9, 2019 (II)]

- (a) 1.0 pm                      (b) 2.0 pm  
(c) 0.1 nm                      (d) 2.0 nm

Ans. (b)

$$\begin{aligned} 7. \quad (b) \quad \text{No. of surfactant molecule} &= 6 \times 10^{23} \times \frac{10}{1000} \times 10^{-3} \\ &= 6 \times 10^{18} \text{ molecule} \end{aligned}$$

Let edge length =  $a \text{ cm}$

$$\begin{aligned} \text{Total surface area of surfactant} &= 6 \times 10^{18} a^2 \\ &= 0.24 \text{ cm}^2 \end{aligned}$$

$$a^2 = \frac{0.24}{6 \times 10^{18}} = 0.04 \times 10^{-18} = 4 \times 10^{-20}$$

$$a = 2 \times 10^{-10} \text{ cm} = 2 \text{ pm}$$