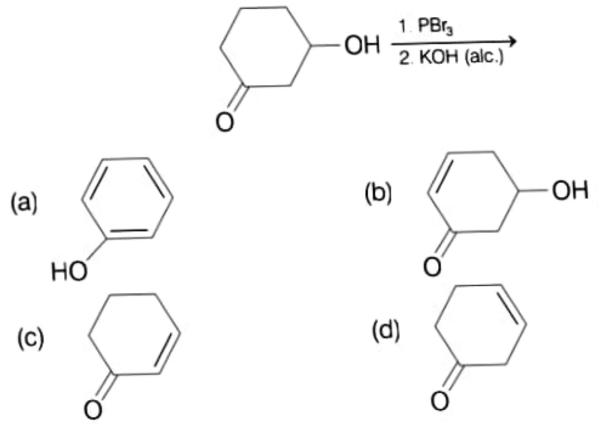
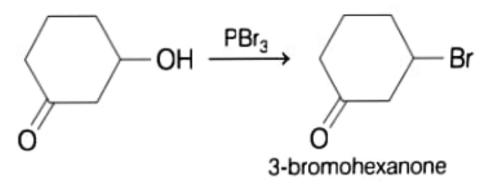
## The major product of the following reaction is (2019 Main, 9 April I)



**Key Idea**  $PBr_3$  reagent is used for the substitution of -Br group while alc. KOH reagent is used to carry out elimination reaction.

The given reaction proceed in following manner:

**Step I** In presence of PBr<sub>3</sub>, alcohols undergo substitution reactions to give halides. Reagent PBr<sub>3</sub> is usually generated *insitu* by the reaction of red phosphorus with bromine.



Step II 3-bromohexanone in presence of alc. KOH undergoes elimination reaction and gives cyclohex-2-en-1-one.

