Exemplar Problems

Q5. Which of the carbonates given below is unstable in air and is kept in $C0_2$ atmosphere to avoid decomposition.

- (a) BeCO₃
- (b) MgC0₃
- (c) CaCO₃
- (d) BaCO₃

Sol: (a) Due to least electropositive character or least basicity of Be, BeC0₃ is less stable and hence decomposes to give BeO and C0₂.

BeC0₃ \rightarrow BeO + C0₂

Since the decomposition reaction is reversible, therefore, to increase the stability of $BeCO_3$ or to reverse the above equilibrium, $BeCO_3$ is kept in an atmosphere of CO_2 .