## **Exemplar Problems**

Q4. Metal carbonates decompose on heating to give metal oxide and carbon dioxide. Which of the metal carbonates is most stable thermally?

- (a)  $MgCO_3$
- (b)CaCO<sub>3</sub>
- (c)SrCQ<sub>3</sub>
- (d)BaC0<sub>3</sub>

Sol: (d) Thermal stability of metal carbonates increases as the electropositive character of the metal or the basicity of the metal hydroxide increases from  $Be(OH)_2$  to  $Ba(OH)_2$ . Thus,  $BaCO_3$  is the most stable.