Exemplar Problems

Q30. Wavelengths of different radiations are given below:

$$\lambda(A) = 300 \text{ nm}, \ \lambda(B) = 300 \ \mu\text{m}, \ \lambda(C) = 3 \text{ nm}, \ \lambda(D) = 30 \ \text{Å}$$

Arrange these radiations in the increasing order of their energies.

Sol.
$$E = hv$$
 or $= \frac{hc}{\lambda}$ or $E \propto \frac{1}{\lambda}$
 $\lambda(A) = 300 \text{ nm} = 300 \times 10^{-9} \text{ m or} = 3 \times 10^{-7} \text{ m};$
 $\lambda(B) = 300 \times 10^{-6} \text{ m} = 3 \times 10^{-4} \text{ m}$
 $\lambda(C) = 3 \times 10^{-9} \text{ m}, \ \lambda(D) = 30 \times 10^{-10} \text{ m} = 3 \times 10^{-9} \text{ m}$
 \therefore Increasing order of energies is:
 $B < A < C = D$