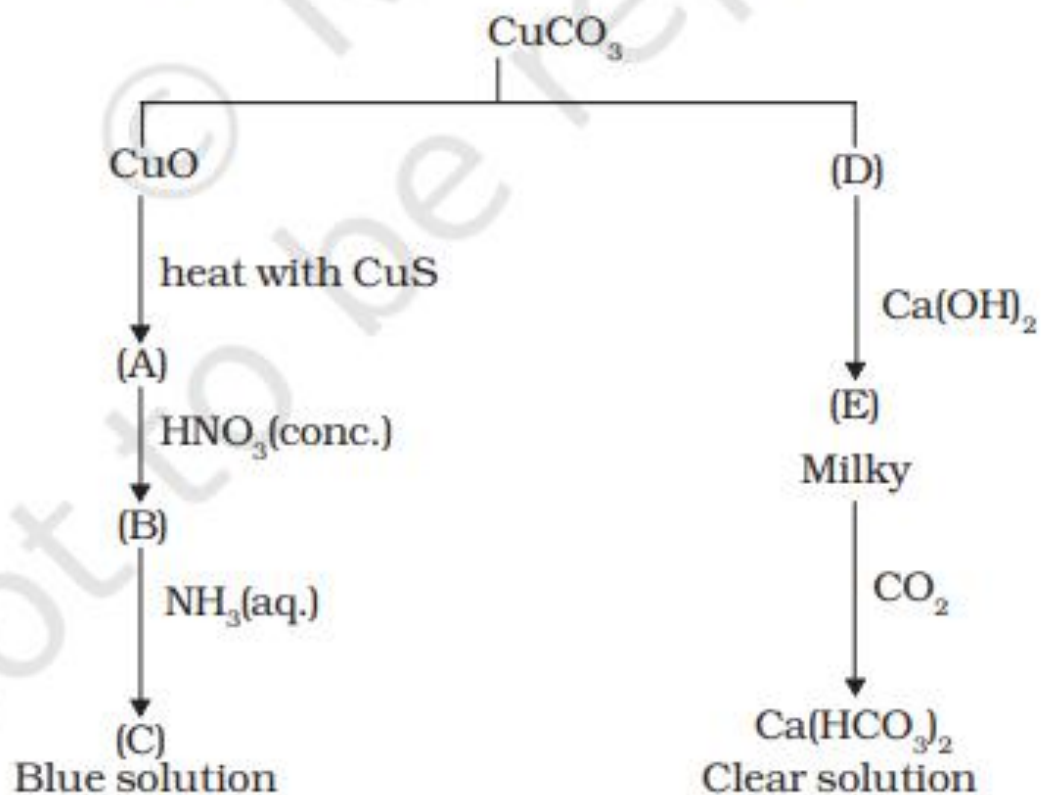


64. Identify A to E and also explain the reactions involved.



64. A = Cu B = $\text{Cu}(\text{NO}_3)_2$ C = $[\text{Cu}(\text{NH}_3)_4]$ D = CO_2

E = CaCO_3 F = $\text{Cu}_2[\text{Fe}(\text{CN})_6]$ G = $\text{Ca}(\text{HCO}_3)_2$

