32.	Why does copper not replace hydrogen from acids?
32. Cu sl	nows positive E value.

38. When a brown compound of manganese (A) is treated with HCl it gives a gas (B). The gas taken in excess, reacts with NH₃ to give an explosive compound (C). Identify compounds A, B and C.

38. A = MnO2 B = Cl2 C = NCl3

MnO2 + 4HCl
$$\longrightarrow$$
 MnCl2 + Cl2 + 2H2O (A) (B)

NH3 + 3Cl2 \longrightarrow NCl3 + 3HCl (excess) (C)

45. When orange solution containing $Cr_2O_7^{-2}$ ion is treated with an alkali, a yellow solution is formed and when H $^+$ ions are added to yellow solution, an orange solution is obtained. Explain why does this happen?

45.
$$\operatorname{Cr_2O_7^{2^-}} \xrightarrow{\operatorname{OH^-}} \operatorname{CrO_4^{2^-}}$$
Dichromate Chromate (Orange) (Yellow)