Related Problems with Solutions

Problem 2:

9.28 How many ions are produced from the complex $Co(NH_3)_6Cl_2$ in solution?(i) 6(ii) 4(iii) 3(iv) 2

Solution:

Ans: Coordination number of cobalt = 6

Hence, the complex is $[{\rm Co}~({\rm NH}_3)_6]~{\rm Cl}_2.$ It ionizes in the solution as follows

$[\operatorname{Co}(\operatorname{NH}_3)_6]\operatorname{Cl}_2 \xrightarrow{+aq} [\operatorname{Co}(\operatorname{NH}_3)_6]^{2+} + 2\operatorname{Cl}^{-}$

Thus, three ions are produced. Hence, the correct option is (iii)