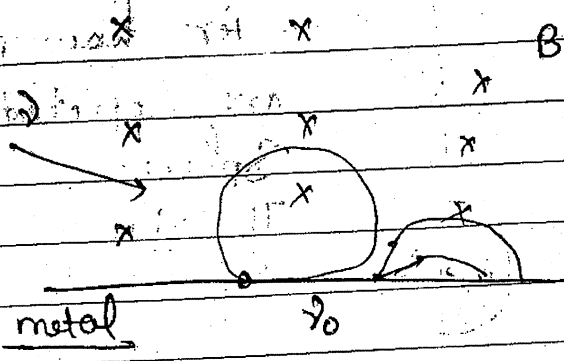


Q-3



what maximum distance of an electron from metal surface after ejection.

$$r_{\max} = \frac{\sqrt{2m E_{\max}}}{eB}$$

$$E_{\max} = h\nu - h\nu_0$$

$$2r_{\max} = \frac{2\sqrt{2m E_{\max}}}{eB}$$

De-broglie Wavelength

As the light shows dual nature of wave and particle together similarly the particles, also shows dual nature.

The wave associated with particle matter is called - matter wave.

Wavelength of ~~metal~~ matter wave is given by

$$\lambda = \frac{h}{p}$$

$$= \frac{h}{mv} \text{ - Speed}$$

$$= \frac{h}{\sqrt{2mE}}$$

$\lambda = \frac{h}{p} = \frac{h}{mv}$ where $v = \sqrt{2mqV}$ (voltage in V)
 $K.E = qV \rightarrow \text{BD}$

$$\lambda = \frac{h}{p} = \frac{h}{mv_{\text{speed}}} = \frac{h}{\sqrt{2mE}} = \frac{h}{\sqrt{2mqV}}$$

Q1. If proton and α -particle are accelerated through same P.D. what is ratio of their de Broglie wavelength.

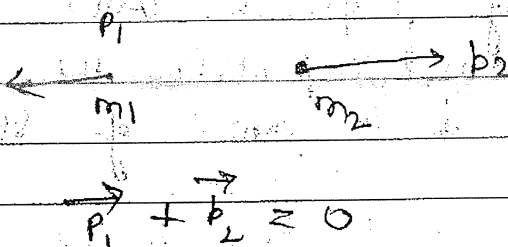
Ans $\lambda = \frac{h}{\sqrt{2mqV}}$

$$\lambda_p = \frac{h}{\sqrt{2 \times 1 \times 1 \times V}}$$

$$\lambda_\alpha = \frac{h}{\sqrt{2 \times 4 \times 2 \times V}}$$

$$\frac{\lambda_p}{\lambda_\alpha} = \frac{2\sqrt{2}}{1}$$

Rest

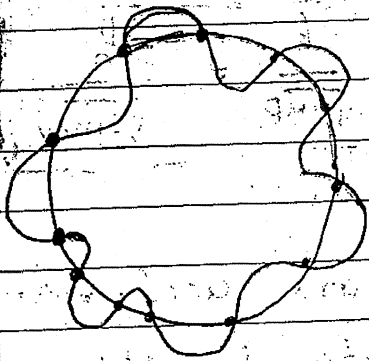


$$|\vec{p}_1| = |\vec{p}_2|$$

$$\lambda_1 = \lambda_2$$

Momentum not same due to direction because momentum is (vector).

If an electrons revolves in Bohr's orbit its circumference is n times of de-broglie wavelength where n is principal quantum number. $= 3 \times \lambda$



$$mvr = \frac{nh}{2\pi}$$

$$2\pi r = n \lambda$$

$$\text{Circumference} = n \lambda$$

$$\text{Circumference} = n \lambda$$

Wave function of the wave associated with any particle represents the probability of finding of particle at a place or (position)

Place where wave function is zero, electron finding in zero.

For free particle the wave function is in form of progressive wave.

while for a bound particle the wave function is in form of standing wave.

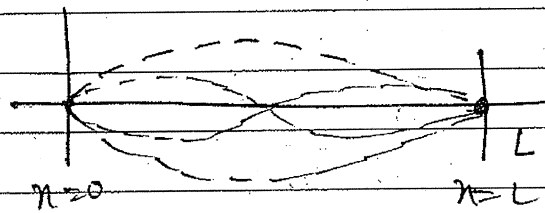
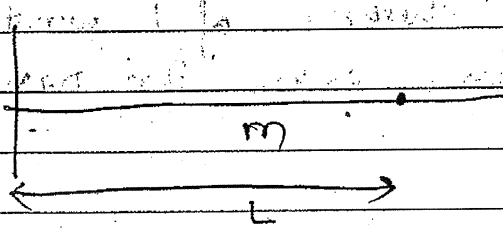
(vector) \vec{a}

$$0 = \vec{a} + \vec{a}$$

$$|\vec{a}| = |\vec{a}|$$

$$k = k$$

Q1) A particle of mass m is bound to exist b/w $x=0$ & $x=L$, find possible value of energy of particle. Find the minimum energy that particle can be observed.



$$L = n \frac{\lambda}{2}$$

$$\lambda = \frac{2L}{n}$$

$$\frac{h}{p} = \frac{2L}{n}$$

$$p = \frac{nh}{2L}$$

$$E_n = \frac{p^2}{2m} = \frac{n^2 h^2}{8L^2 m}$$

$$(\Delta E)_{\min} = E_2 - E_1$$

$$= \frac{3h^2}{8mL^2}$$

Q2) If two particles are moving mutually \perp such that their de Broglie wavelength are λ_1 and λ_2 respectively.

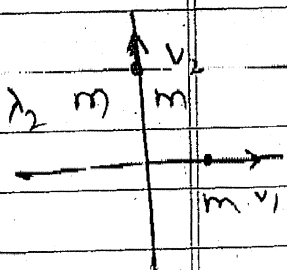
(i) Find de Broglie wavelength of 1 w.r.t to other

(ii) Find de Broglie wavelength of 1 w.r.t to centre of mass (assume that masses are equal.)



$$\lambda_1 = \frac{h}{mv_1} \quad \lambda_2 = \frac{h}{mv_2}$$

$$\lambda_{1/2} = \frac{h}{m v_{1/2}} = \frac{h}{m \sqrt{v_1^2 + v_2^2}}$$



$$m = \frac{h}{\sqrt{\frac{h^2}{m^2 \lambda_1^2} + \frac{h^2}{m^2 \lambda_2^2}}}$$

$$\vec{v}_{cm} = \frac{v_1 \hat{i} + v_2 \hat{j}}{2}$$

w.r.t m

$$\vec{p}_1 + \vec{p}_2 = 0$$

$$\vec{p}_1 = -\vec{p}_2$$

$$\lambda_{1/cm} = \frac{h}{m |\vec{v}_{1/cm}|}$$

$$= \frac{h}{m \sqrt{v_1^2 + v_2^2}}$$

done
 2.70
 wavelength
 relative to
 centre of mass
 is given by
 $\lambda_{1/cm} = \frac{h}{m \sqrt{v_1^2 + v_2^2}}$
 where $v_{1/cm}$ is the velocity of the particle relative to the centre of mass.