

Temperature Coefficient ( $Q$ ) =  $\frac{\text{Rate of Rxn. at } (x+10)^{\circ}\text{C}}{\text{Rate of Rxn. at } x^{\circ}\text{C}} \approx 2$  (Generally)

$\Rightarrow$  Rate of reaction doubles for every  $10^{\circ}$  rise in Temperature

Exact Expression is given by Arrhenius Equation:

$$k = A e^{-E_a/RT}$$