Mass Defect

Mathematically :- ΔM = [Z m_p+ (A-Z) m_n]- M

 Where m_p=mass of 1 proton, Z=number of protons,(A-Z)= mass of neutrons, m_N = mass of 1 neutronand M =nuclearmass of the atom.

Binding Energy

- $\circ E_b = \Delta Mc^2$
 - Where E_h = nuclear binding energy, ΔM =mass defect.