

What is the Fajan's rule about?

- a) electronegativity
- b) ionic compounds
- c) Oxidation State
- d) covalent compounds

Answer: c

Explanation: The Fajan's rule is that the compounds in +2 oxidation state are ionic in nature and +4 oxidation state is covalent in nature, therefore the Fajan's rule is about Oxidation state and their nature of the compounds.